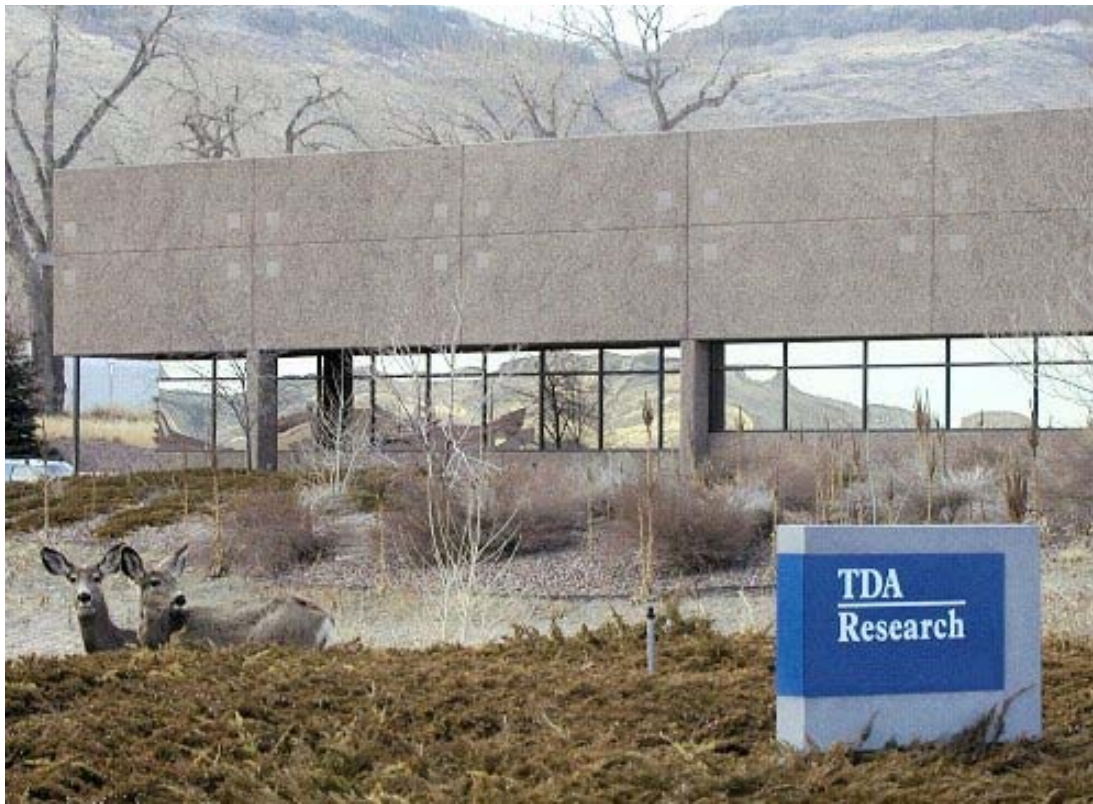


# **A Low Cost, High Capacity Regenerable Sorbent for Pre-Combustion CO<sub>2</sub> Capture**

**Contract No. DE-FE0000469**

**Project Briefing**



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**NETL CO<sub>2</sub> Capture  
Technology Meeting**

**Pittsburgh, PA  
July 12, 2012**

**TDA Research Inc. • Wheat Ridge, CO 80033 • [www.tda.com](http://www.tda.com)**

# Project Objective

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- **The objective is to develop a new pre-combustion carbon capture technology and demonstrate its techno-economic viability**
- **A mesoporous carbon grafted with surface functional groups that remove CO<sub>2</sub> via physical adsorption above the dew point of the synthesis gas**
  
- **Budget Period 1**
  - Sorbent optimization and production scale-up
  - Bench-scale evaluations
  - Process design and optimization
  
- **Budget Period 2**
  - Demonstrate sorbent life for 10,000 cycles
  - Slipstream demonstration using actual synthesis gas
  - Conduct an economic analysis to estimate the cost of CO<sub>2</sub> capture

# Project Partners

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**TDA Research**



UNIVERSITY of CALIFORNIA • IRVINE

**ConocoPhillips**

**SOUTHERN  
COMPANY**

**MWV** MEADWESTVACO

## Project Duration

- Start Date = November 15, 2009
- End Date = September 30, 2012

## Budget

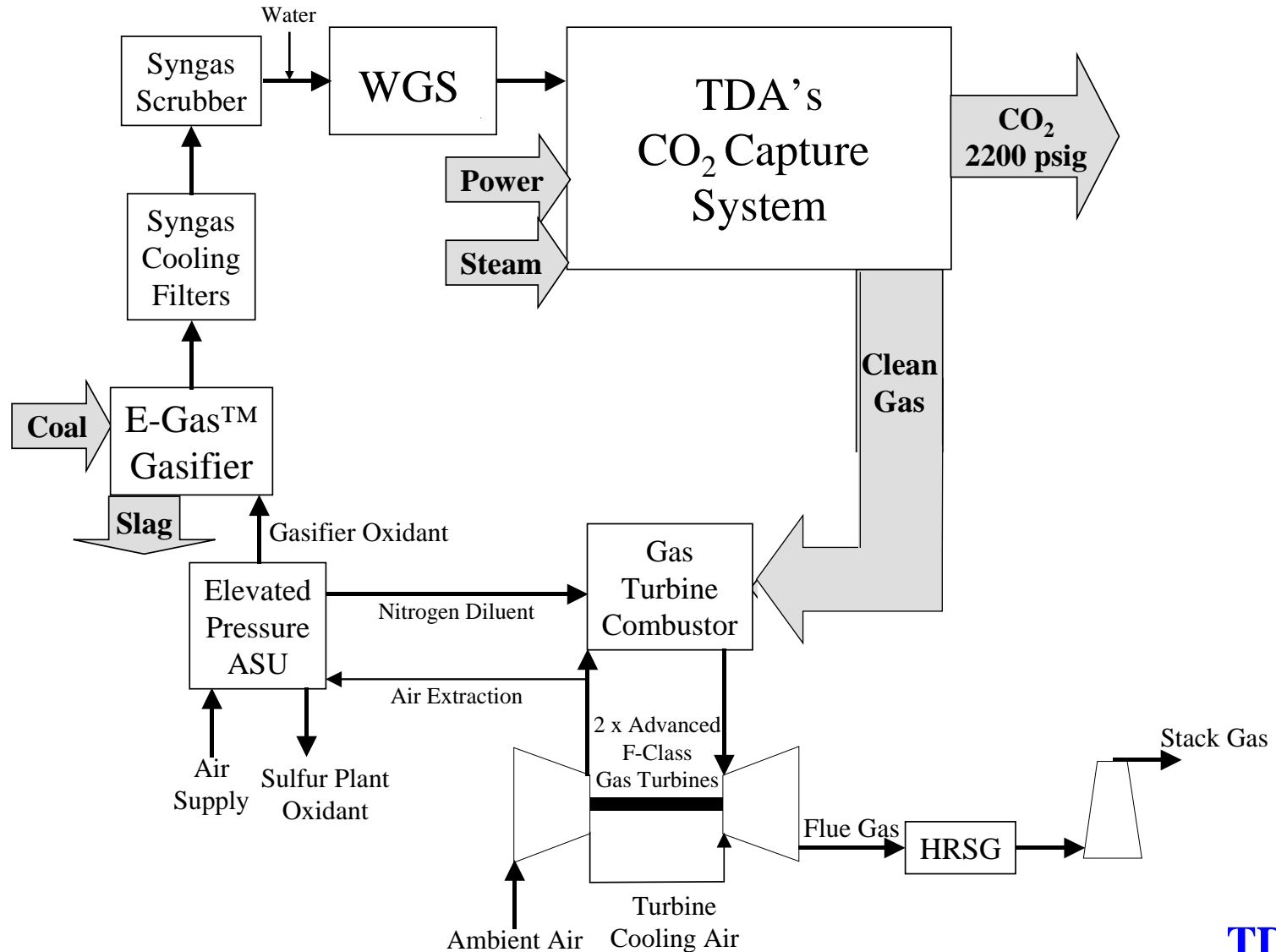
- Project Budget = \$2,500,000
- DOE Share = \$2,000,000
- TDA/Partners Share = \$500,000

# TDA's Approach

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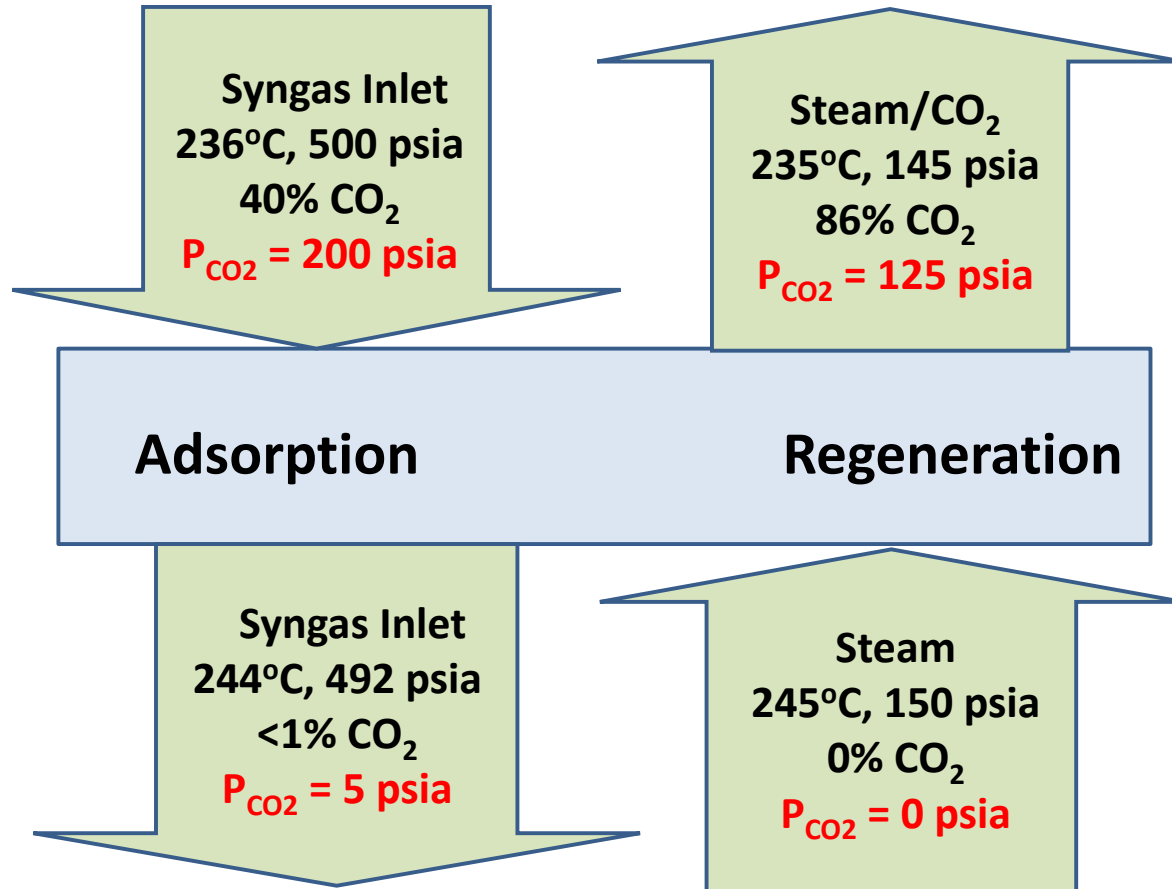
- The sorbent consists of a carbon material modified with surface functional groups that remove CO<sub>2</sub> via strong physical adsorption
  - CO<sub>2</sub>-surface interaction is strong enough to allow high T operation
  - Because there is no covalent bond, energy input for regeneration is low
- Heat of adsorption of CO<sub>2</sub> is **4.9 kcal per mol** for TDA sorbent
  - Selexol ~4 kcal/mol
  - Amine solvents ~14.4 kcal/mol
  - Chemical absorbents 20-40 kcal/mol (Na<sub>2</sub>CO<sub>3</sub> → NaHCO<sub>3</sub> 30 kcal/mol)
    - Na<sub>2</sub>CO<sub>3</sub> + CO<sub>2(g)</sub> + H<sub>2</sub>O<sub>(g)</sub> → 2NaHCO<sub>3</sub> (ΔH = -30 kcal/mol)
- Net energy loss in sorbent regeneration is similar to Selexol
  - A much better IGCC efficiency due to higher temperature CO<sub>2</sub> capture
  - Warm gas clean-up improves cycle efficiency 2 to 4%
  - Potential for even higher efficiency by integrating WGS/CO<sub>2</sub> removal processes (3 to 6% improved efficiency)

# IGCC-Integrated CO<sub>2</sub> Capture System



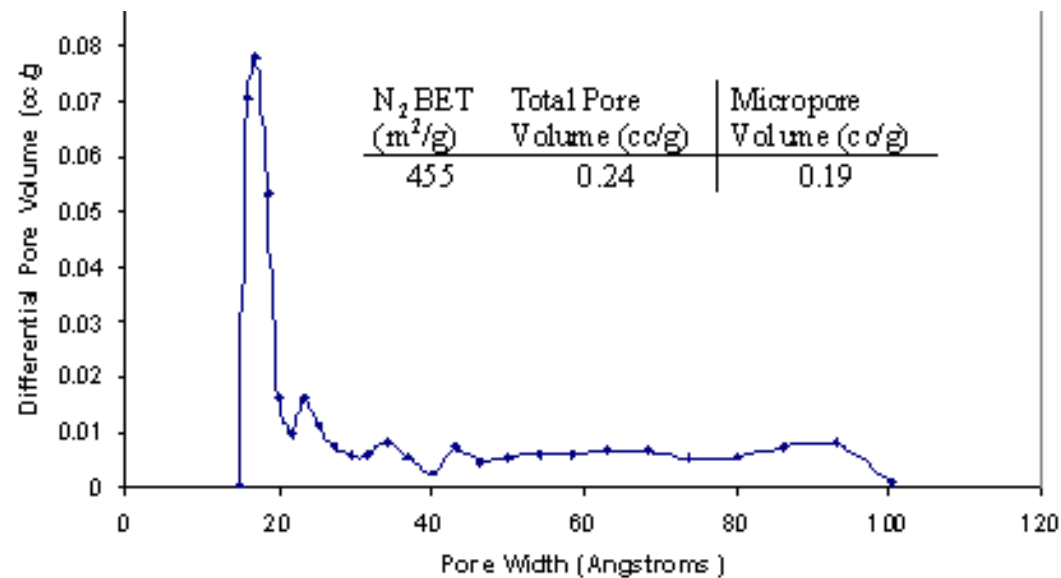
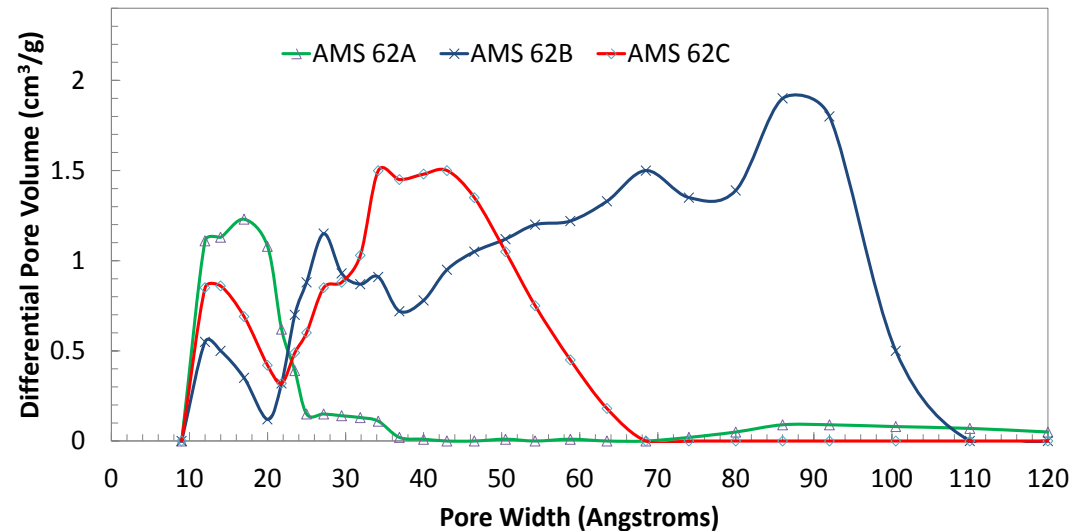
# Regeneration Options

- **Physical adsorbent provides flexibility in regeneration**
  - Temperature swing
  - Pressure swing
  - Concentration swing
  - Any combinations
- **Isothermal operation is critical to eliminate heat/cool transitions which reduces cycle time and increases sorbent utilization**
- **Steam consumption can be reduced significantly if steam purge is carried out at low pressure**



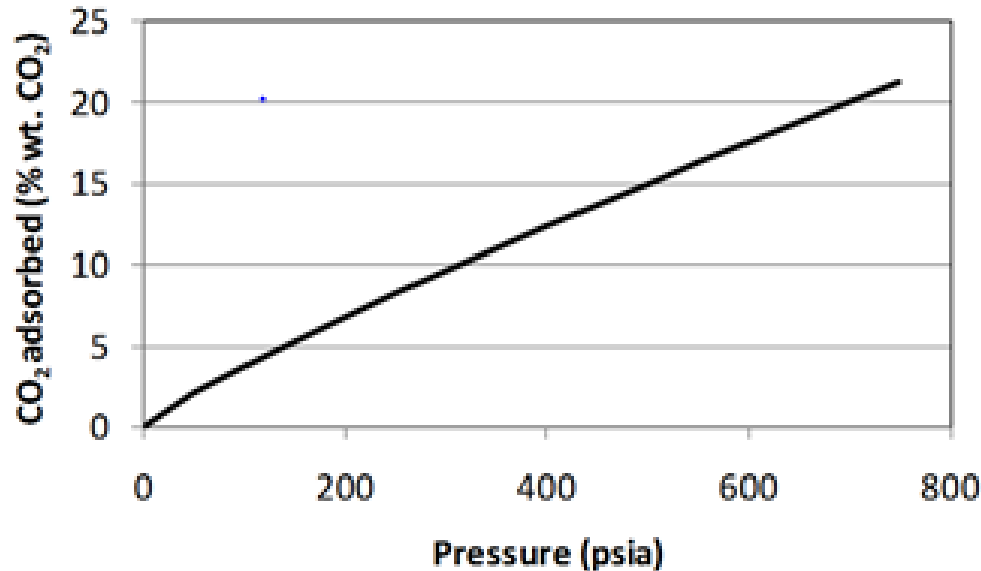
# TDA's Sorbent

- A mesoporous carbon is used to disperse the active sorbent phase
- The carbon support is previously developed for ultra-capacitors, large pores to achieve liquid transport
- The preparation process enables us to introduce various surface groups active for removing different compounds
- Large pores accommodate different active groups



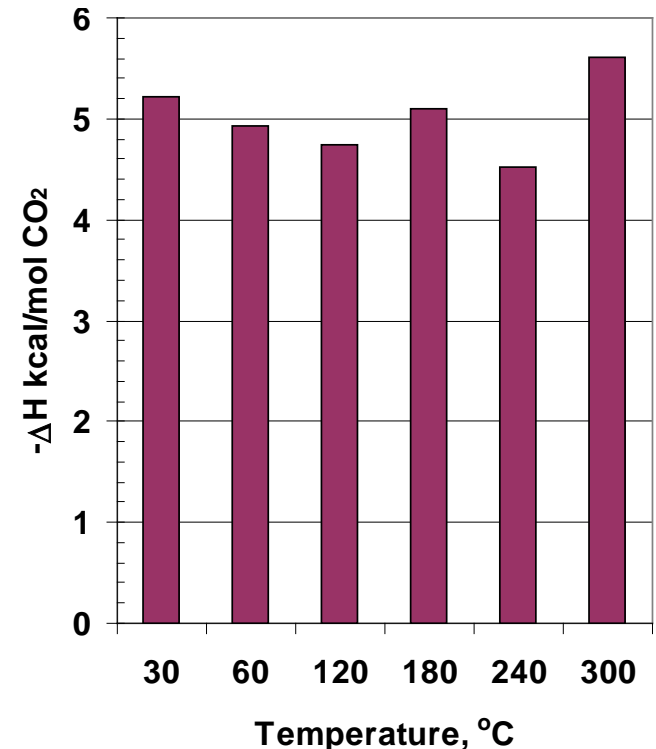
# CO<sub>2</sub> Isotherm and Heat of Adsorption

CO<sub>2</sub> isotherm at 240°C



Langmuir Coefficient ( $q_s$ )	386.4 mmol/g
Langmuir Coefficient (B)	4.15E-04 1/atm
Langmuir Coefficient (n)	0.869
Diffusion Coefficient ( $D/R^2$ )	1.32E-03 1/s
Reference Temperature for B	240 °C
Heat of Adsorption ( $\Delta H$ )	4.8 kcal/mol

Calorimetry Measurements



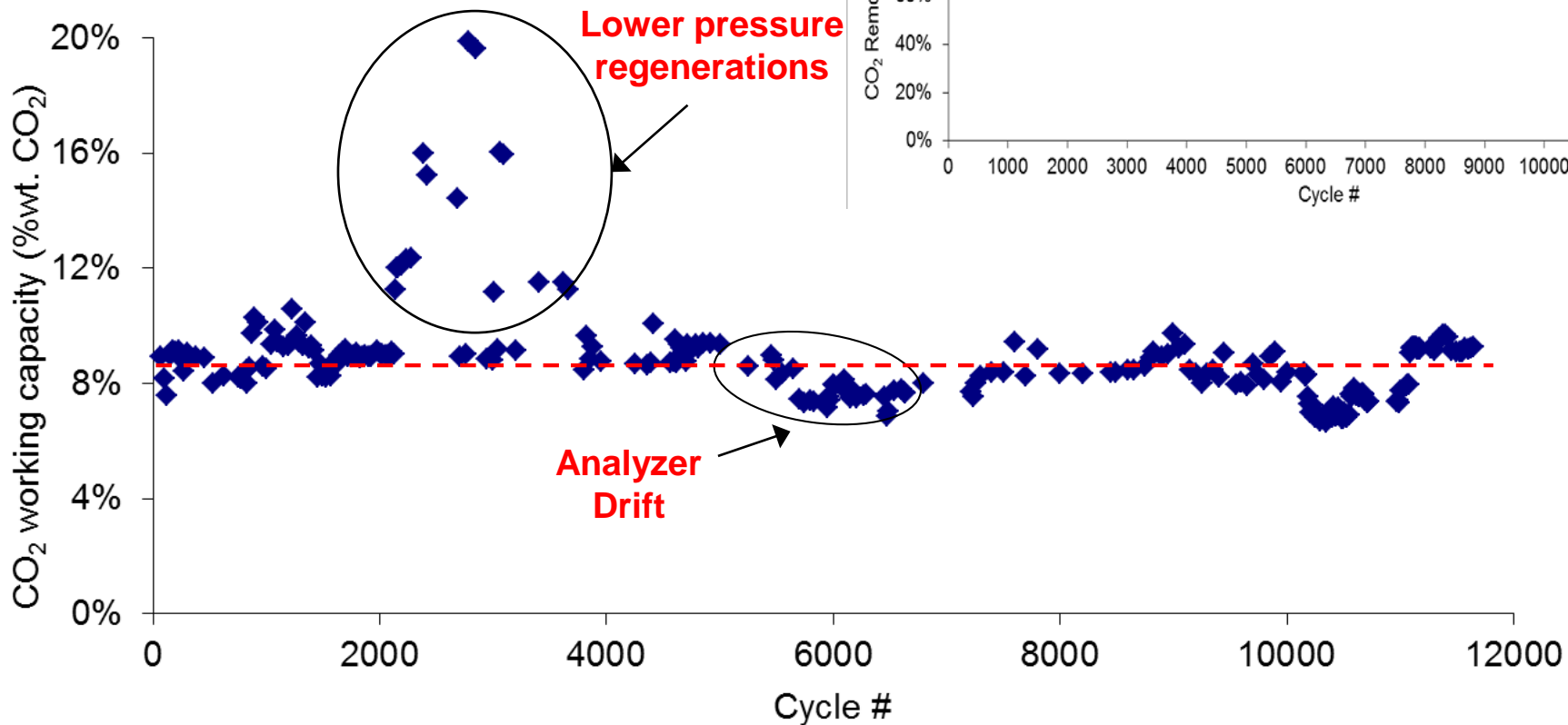
$$-\Delta H_{\text{ads}} = 4.9 \pm 0.4 \text{ kcal/mol}$$

- Isostatic heat of adsorption calculations and DSC experiments confirm the low heat of adsorption



# Multiple Cycle Tests

$H_2=32\%$ ,  $CO_2=40\%$ ,  $N_2=3\%$ ,  
 $CO=1\%$ ,  $H_2O=24\%$ ;  $T= 240^\circ C$ ;  
 $P_{ads}= 500$  psig;  $P_{des} = 50-300$  psig

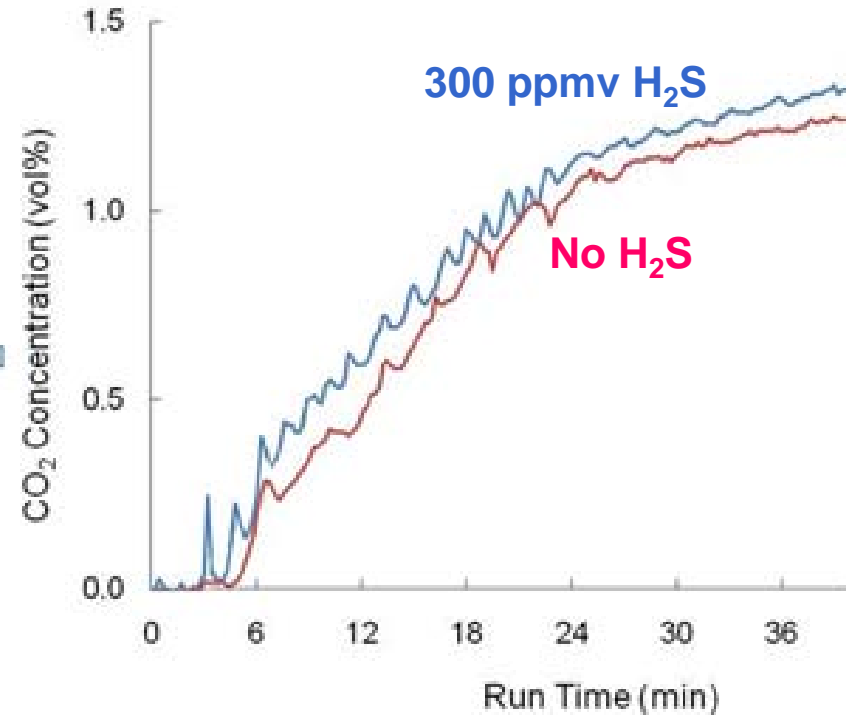
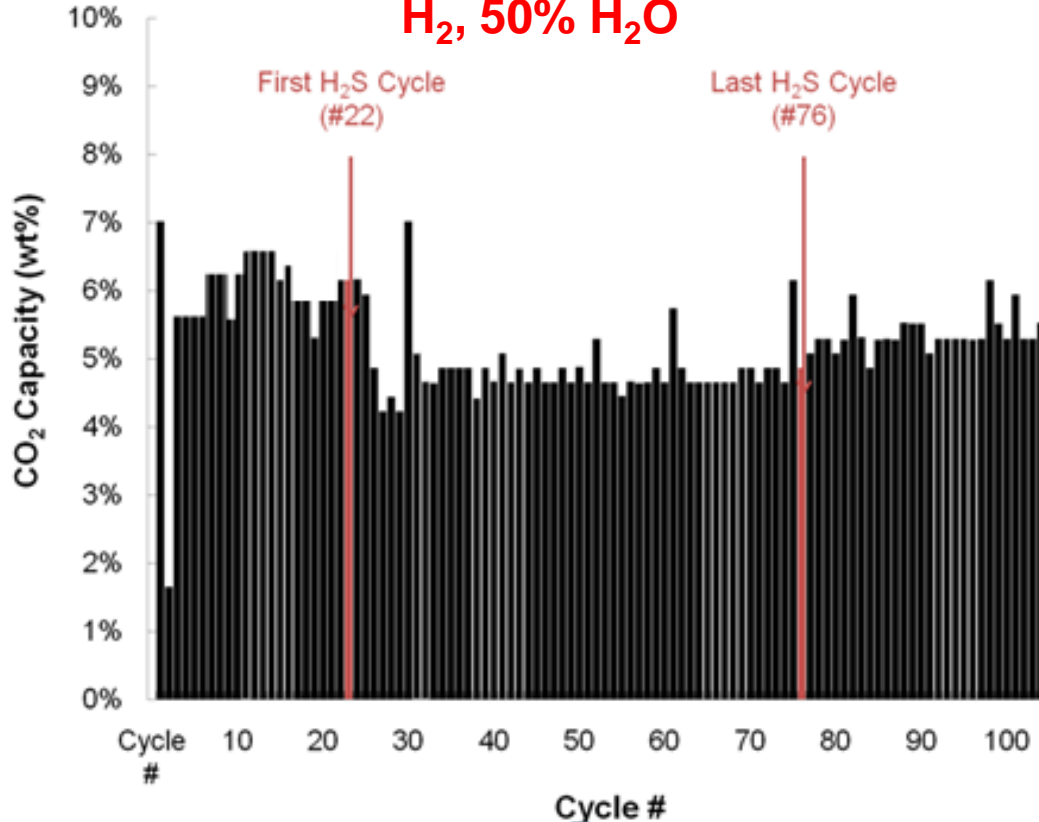


- Sorbent maintained its CO<sub>2</sub> capacity (8+%wt.) for ~12,000 cycles

# Impact of Sulfur

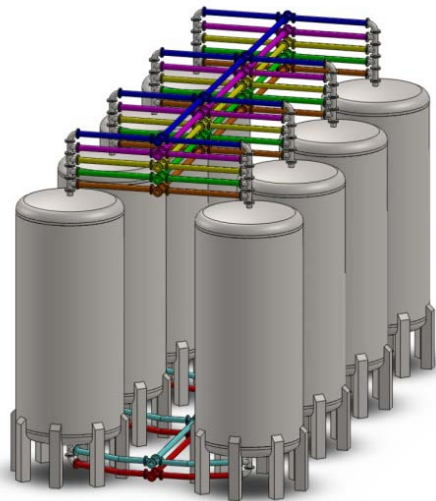
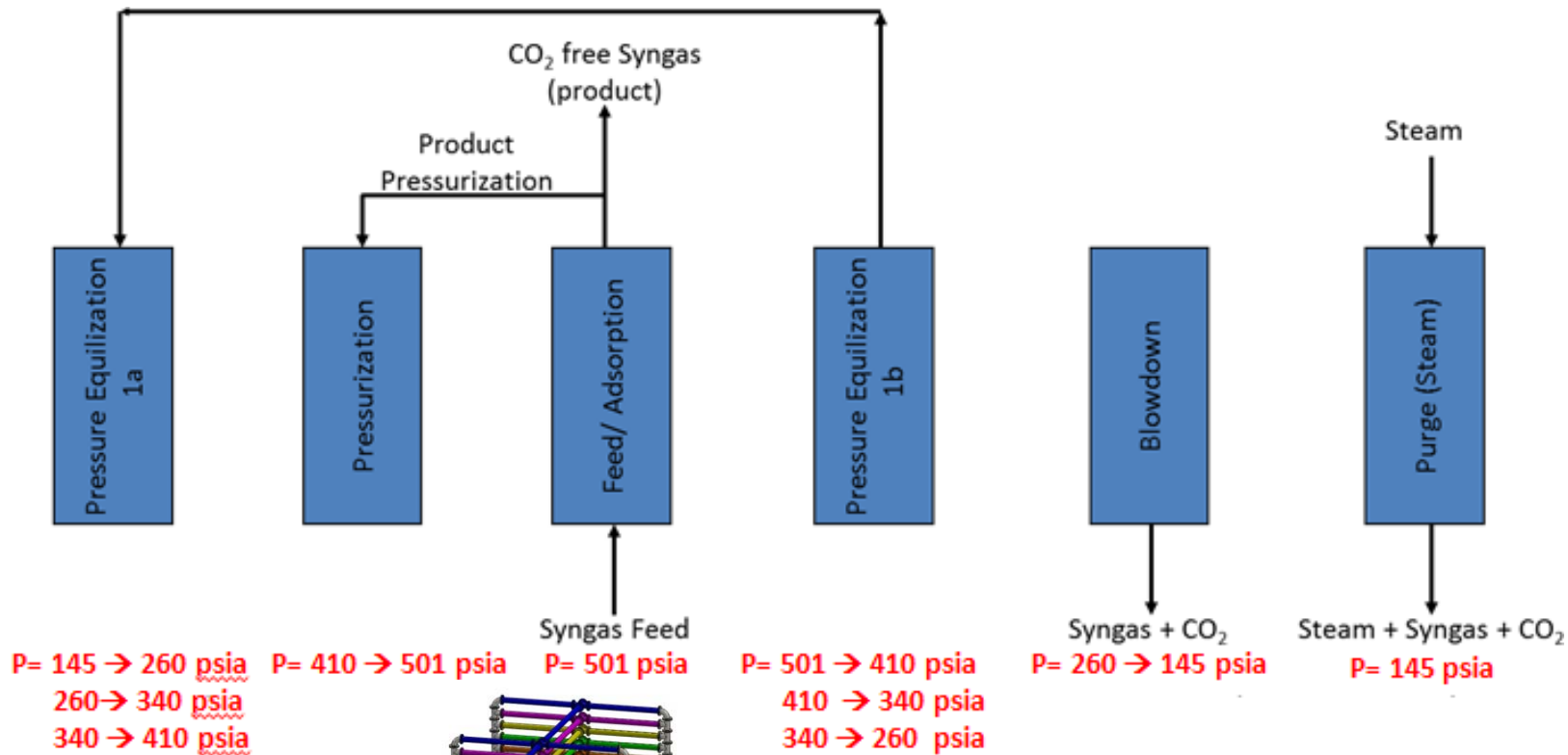
**T= 240°C, P= 500 psig, 10 ppmv H<sub>2</sub>S, 44% CO<sub>2</sub>, 20% H<sub>2</sub>, 36% H<sub>2</sub>O; Purge Gas: 50% H<sub>2</sub>, 50% H<sub>2</sub>O**

**300 ppmv H<sub>2</sub>S, T= 240°C, P= 500 psig**

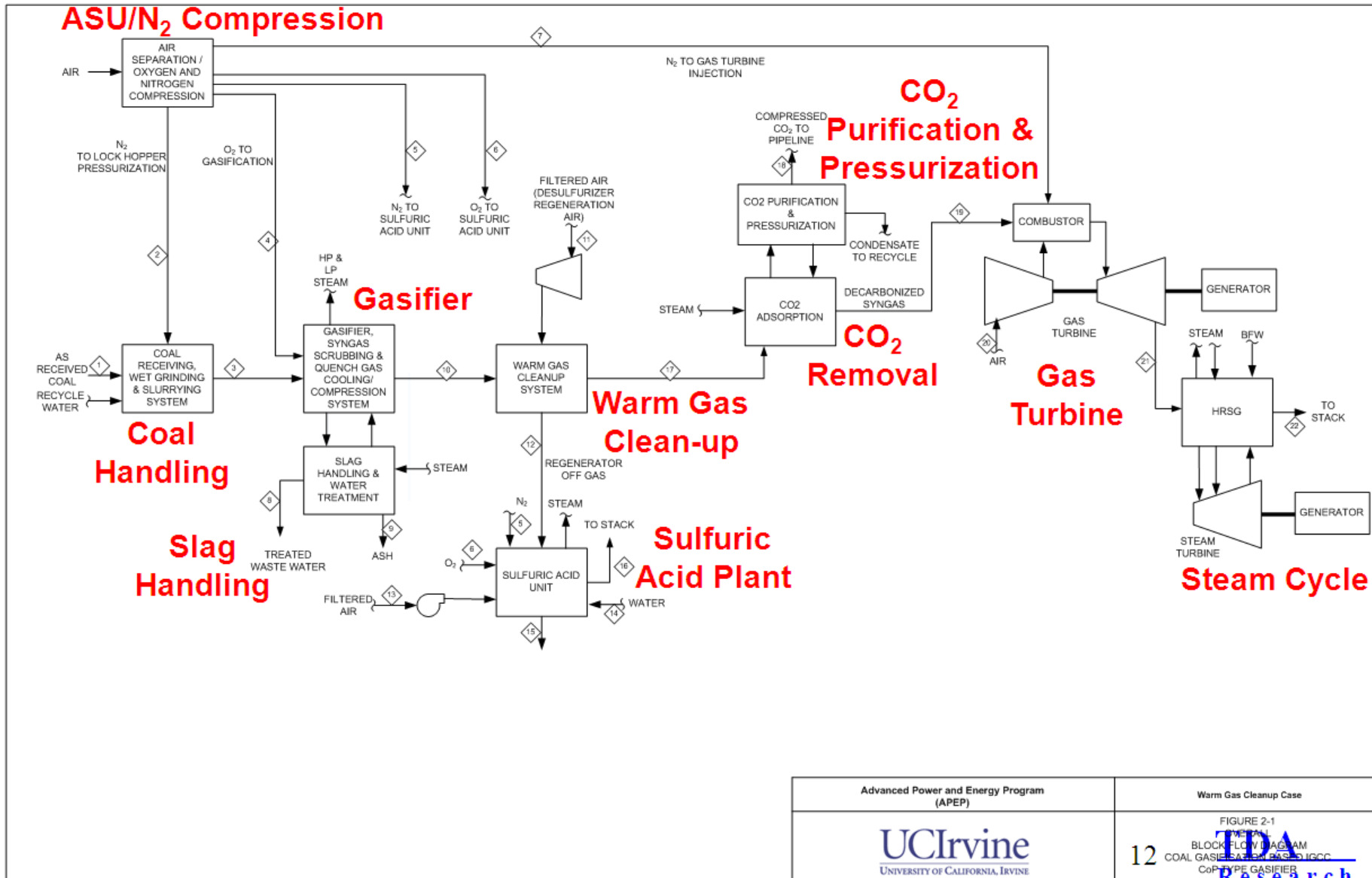


- Presence of H<sub>2</sub>S did not have a significant impact on sorbent performance

# PSA Cycle Design

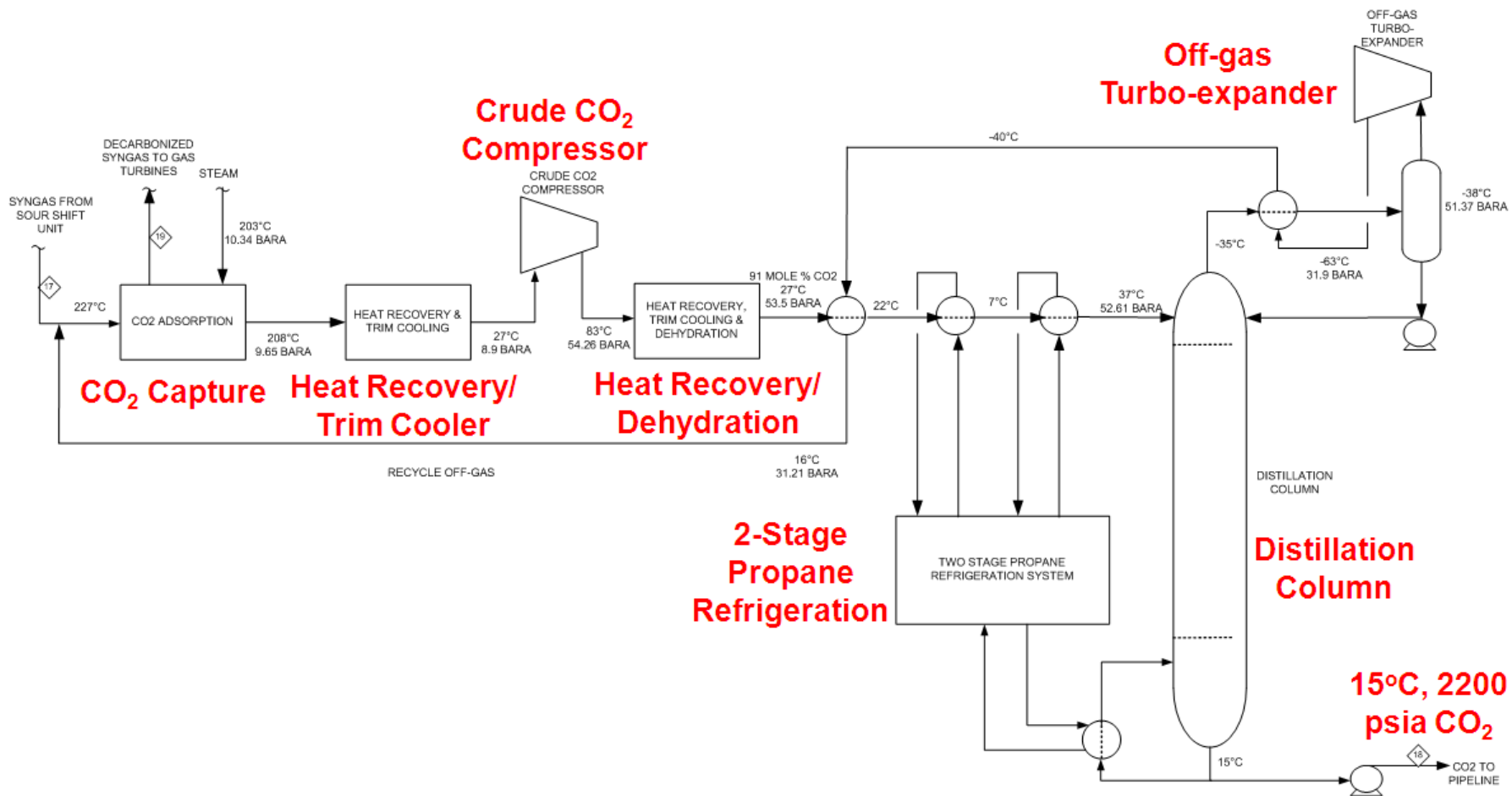


# Process Design and Modeling



Advanced Power and Energy Program (APEP)	Warm Gas Cleanup Case
	FIGURE 2-1 BLOCK FLOW DIAGRAM COAL GASIFICATION BASED IGCC CO <sub>2</sub> TYPE GASIFIER 

# CO<sub>2</sub> Purification & Compression



# System Analysis Results

E-Gas Gasifier	IGCC - Selexol	IGCC - Selexol	IGCC -TDA WGC
	Calibration Case	90% Capture	90% Capture
CO <sub>2</sub> Capture	88.2	90	90
Gross Power Generated, kWe	696,770	691,624	711,083
Net Power, kWe	524,772	516,126	572,398
<b>Net Plant Efficiency, % HHV</b>	<b>32.1%</b>	<b>31.6%</b>	<b>33.5%</b>
COE, mills/kWh		98.6	91.0
<b>COE (inc. TS&amp;M), mills/kWh</b>		<b>103.9</b>	<b>96.0</b>

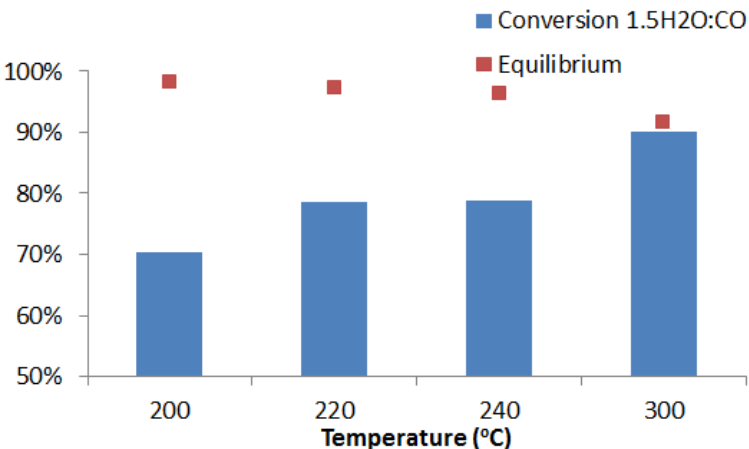
*Model results validated based on DOE/NETL 2007/1281*

- IGCC plant with TDA's CO<sub>2</sub> capture technology system achieves higher efficiency than IGCC-Selexol
  - 33.5% vs. 31.6% for E-Gas gasifier
  - 34.2% vs. 32.4% for GE gasifier
- Cost of electricity with IGCC-TDA-WGC is ~8 mills/kWh lower than that of IGCC-Selexol

# Process Design Modifications

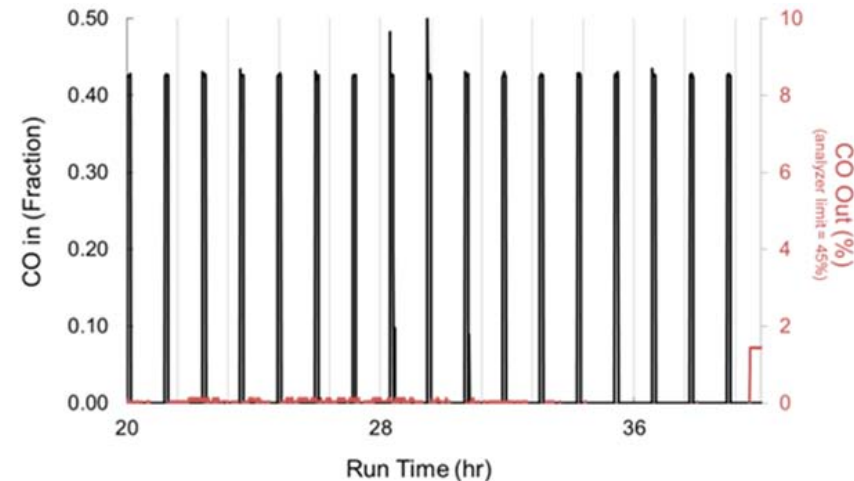
## Integration of 3<sup>rd</sup> stage WGS with the warm gas carbon capture

- Commercial WGS catalyst (Shiftmax 230) co-located with the sorbent
- Cycle efficiency increase to 35+% due to reduced steam consumption
- Lower CAPEX



gas	vol. %
CO <sub>2</sub>	14
CO	37
H <sub>2</sub>	26
N <sub>2</sub>	8
H <sub>2</sub> O	53

T = 240°C; P = 500 psig



- Promising performance data (Advanced Carbon Capture Concept for Low Rank Coals - DE-FE-0007966)
- 7.82% wt. CO<sub>2</sub> loading (5.68% is from conversion of CO into CO<sub>2</sub>)

# Process Design Modifications (Cont'd)

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## Cycle optimization to increase H<sub>2</sub>/syngas recovery (i.e., CO<sub>2</sub> purity)

- Baseline syngas recovery in the sorbent beds = 88%
- Target recovery = 95% (via product purging)
- Lower CAPEX due to reduced cost of purification system



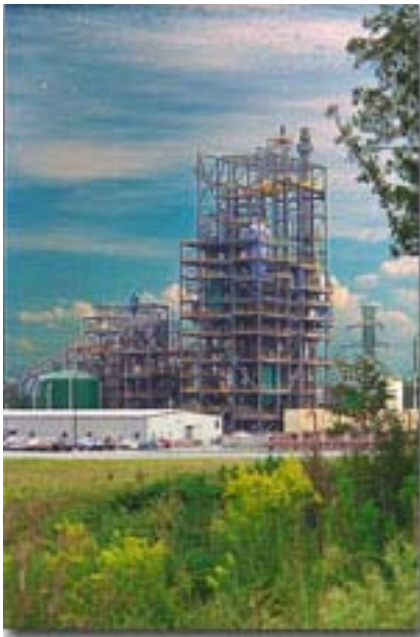
# Slipstream Demonstrations

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- Two 3-week test campaigns for proof-of-concept demonstrations

## Wabash River IGCC Plant, Terre Haute, IN

- Largest single-train gasifier with 262 MW power output
- Oxy-blown E-Gas™ Gasifier
- Operates on petcoke



## National Carbon Capture Center, Wilsonville, AL

- Demonstration in November, 2011
- Pilot-scale gasifier
- Air-blown transport gasifier
- Operates on coals and lignites

# System Pictures – Prior to Insulation

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**Skid #1 - Synthesis gas pre-treatment skid**



**Skid #2 - CO<sub>2</sub> removal skid**

# Test Units – In NEMA-Rated Enclosures

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**CO<sub>2</sub> Removal Skid**

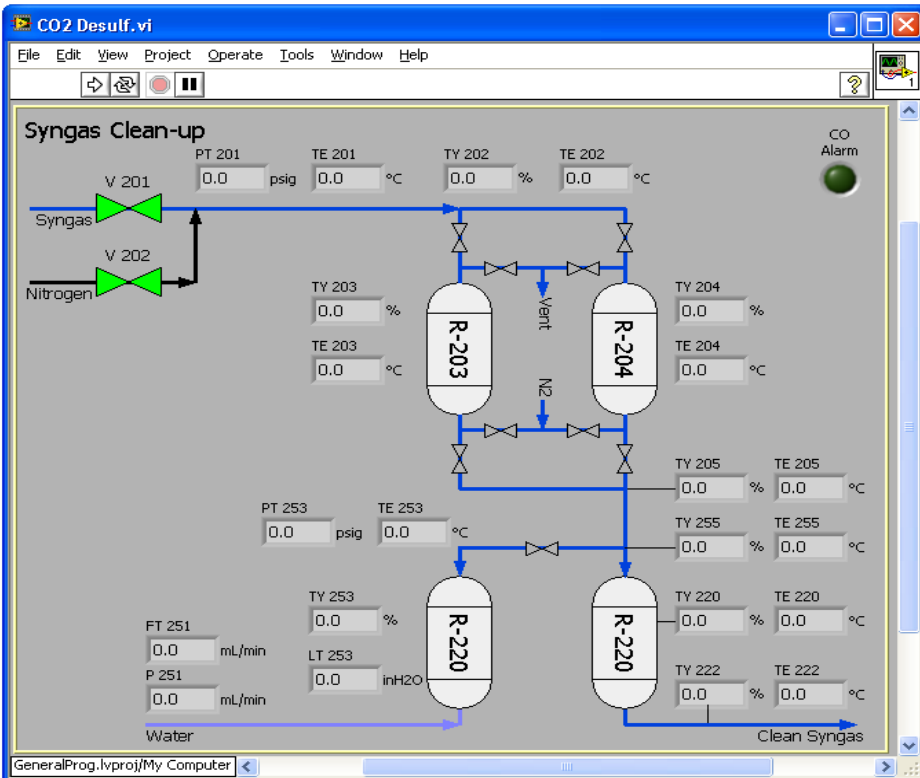


**Gas Conditioning Skid**

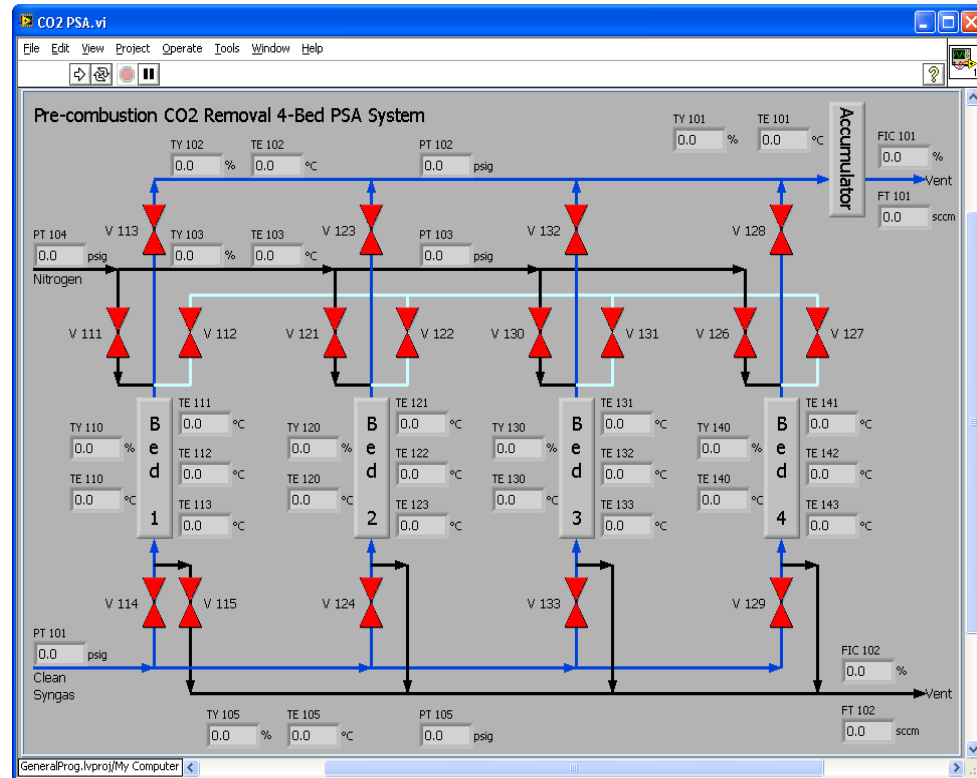


# Control System

## Skid #1



## Skid #2



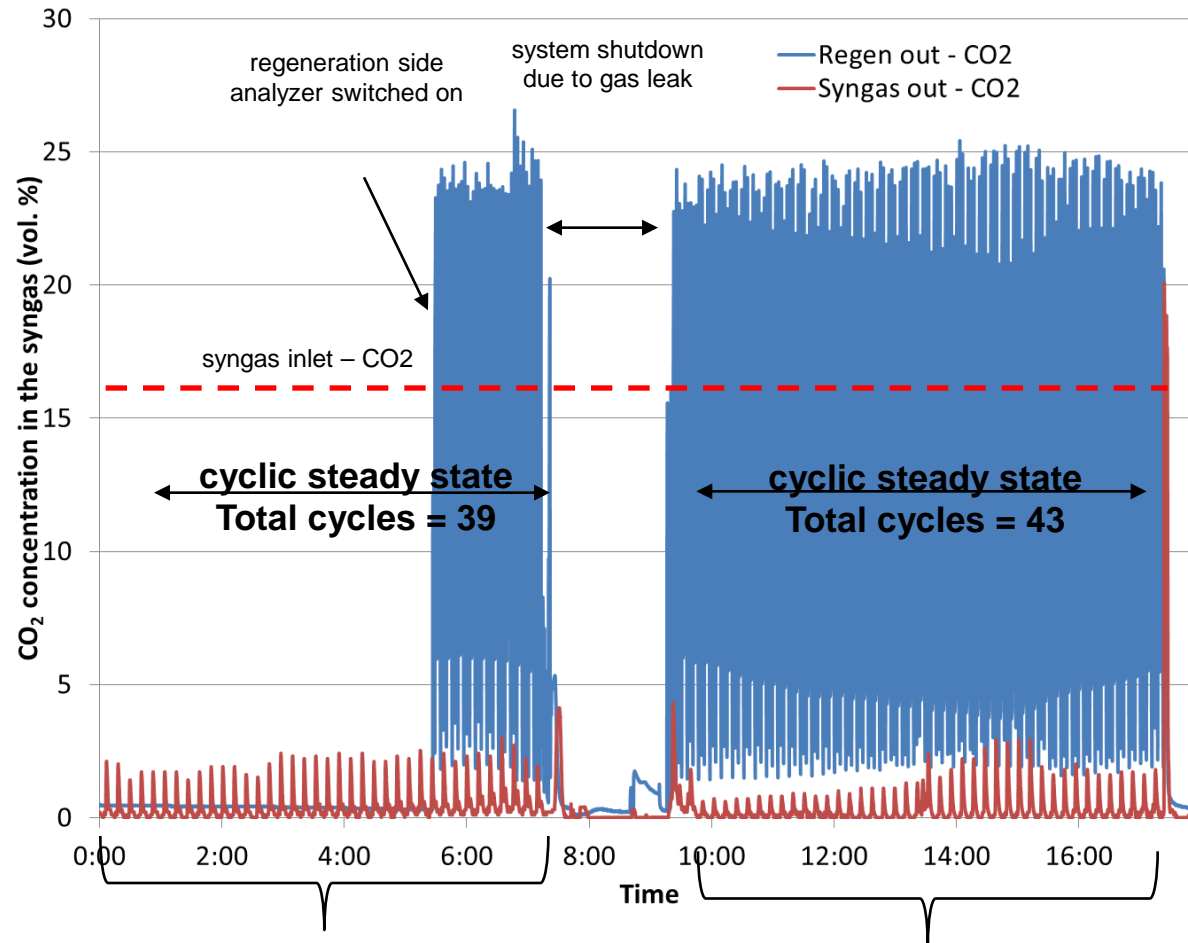
- System is fully automated and remotely controlled

# Field Test Units Installed at Site

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# Field Test Data – November 23 2011

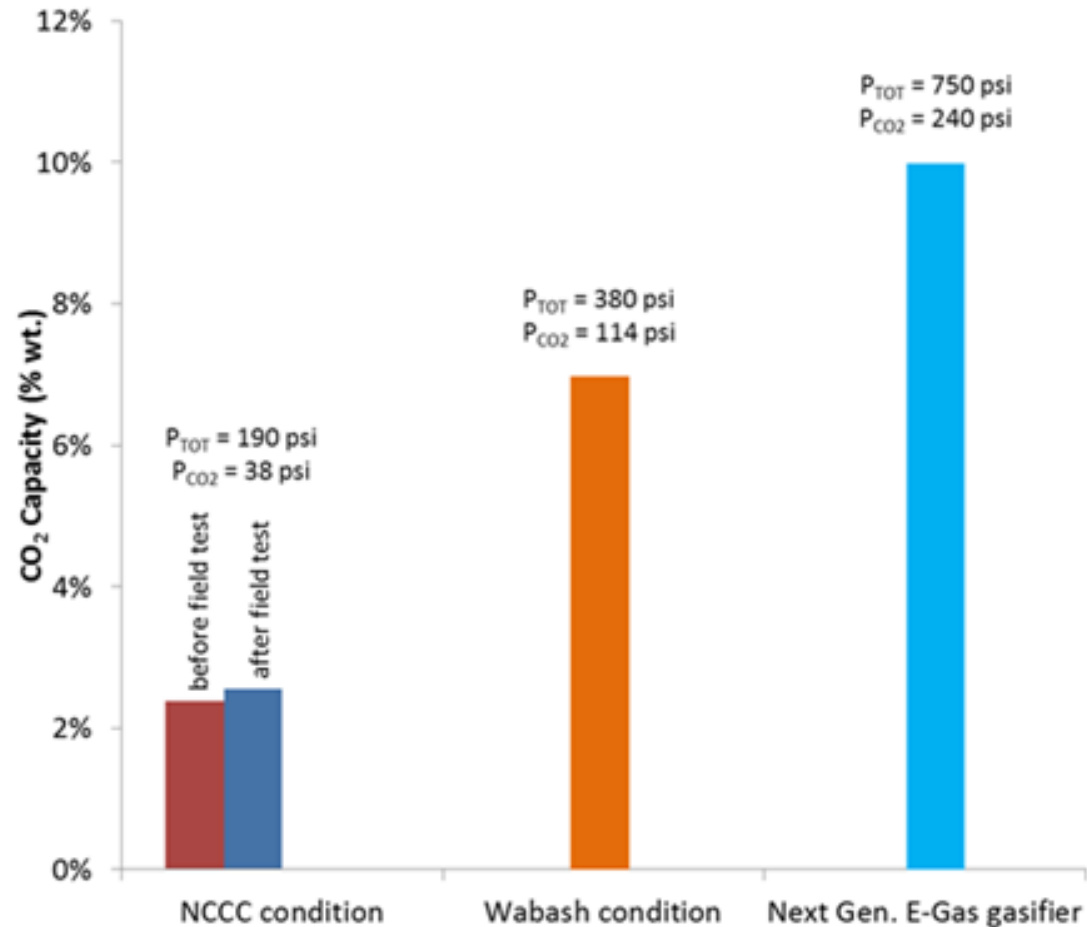


**CO<sub>2</sub> capture ~ 98.3%**  
**Bed Temperature = 229C**  
**Pads = 188 psig; Pdes = 59 psig**

**CO<sub>2</sub> capture ~ 98.7%**  
**Bed Temperature = 226C**  
**Pads = 188 psig; Pdes = 54 psig**

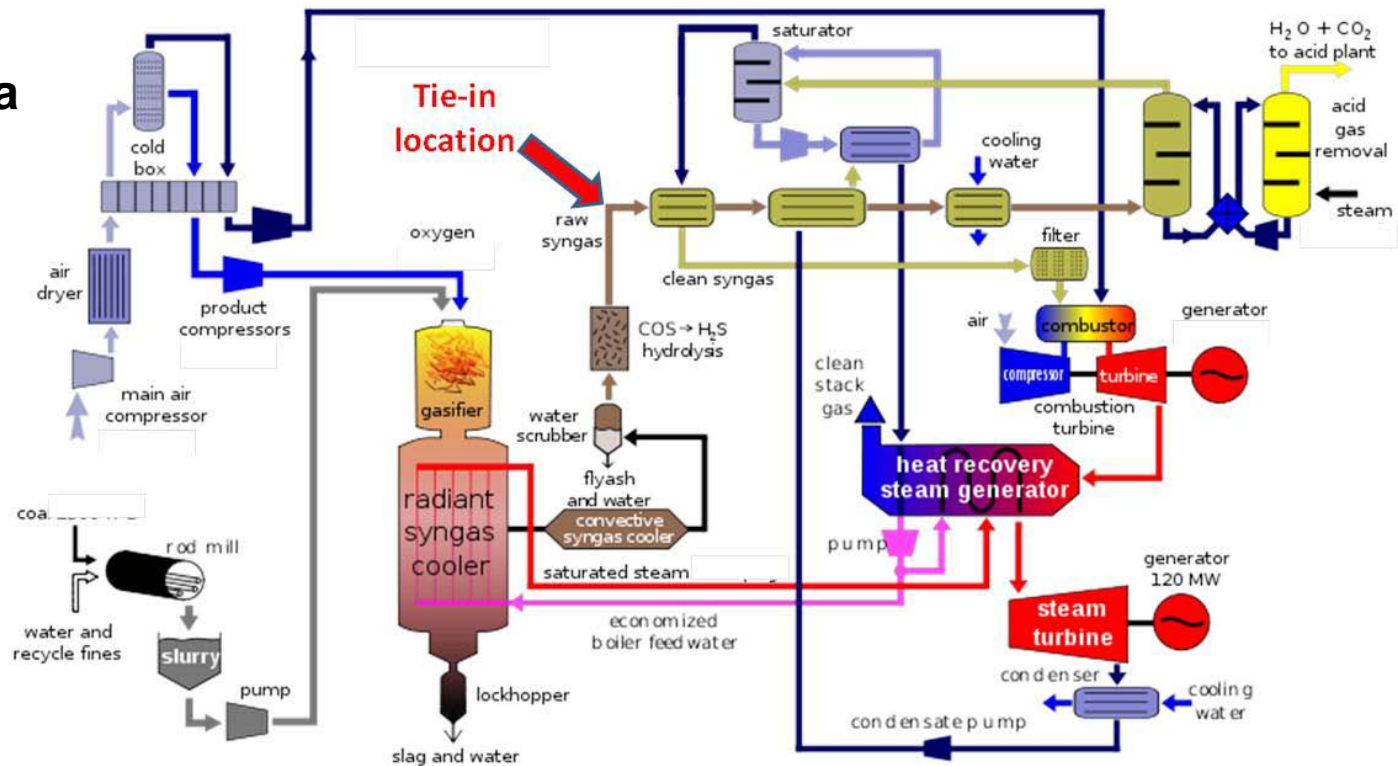
# Performance in Slipstream Tests

- Sorbent maintained a stable capacity for 500 hrs of testing and over 2,000 cycles
- Sorbent maintained CO<sub>2</sub> capacity before and after field test
  - 2.6% wt. CO<sub>2</sub> capacity at P<sub>CO2</sub> = 38 psi
- Sorbent capacity under Wabash condition (P<sub>CO2</sub> = 114 psi) increases to 7.03% wt.
- Next generation E-Gas gasifier is expected to operate at 750 psi (P<sub>CO2</sub> = 240 psi) and the prototype unit achieved 9.94% wt. CO<sub>2</sub> capacity at 750 psi



# Wabash River IGCC Demonstration

- Process Hazard Analysis is complete
- Tie-in location is identified
  - Downstream of a COS hydrolysis unit
  - $T = 194^{\circ}\text{C}$
  - $P = 357 \text{ psia}$



- All site modifications are completed based on TDA's utility requirements
- Scheduled to start in early August 2012



# Acknowledgements

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## **DOE/NETL**

**Arun Bose - DOE Project Manager**

## **Partners**

**Ashok Rao - University of California, Irvine**

**Albert Tsang – Phillips 66**

**Shrinivas Lokare – Phillips 66**

**Frank Morton – Southern Company**

**Tony Wu – Southern Company**

**Paula Walmet – MWV**